Studies with 2-arylhydrazononitriles: further investigations on the utility of 2-arylhydrazononitriles as precursors to 1,2,3-triazole amines

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3-Oxo-2-arylhydrazonobutanenitriles react with hydroxylamine hydrochloride in ethanolic sodium acetate to yield either amidoximes or isoxazolamines depending on the nature of the substituent. The amidoximes cyclise in refluxing DMF containing piperidine to form 2-aryl-1,2,3-triazol-5-amines. The isoxazolamines rearranged into 1,2,3-triazolamines on refluxing in DMF. Amidoxime 2c cyclised in acetic anhydride to give 2-arylhydrazono-1,2,4oxadiazole 9c, which rearranged to acetylamino-1,2,3-triazole 8c. The prepared acetyl-1,2,3-triazolamines 8b,c were utilised as precursors to triazolopyridines and pyrazolyltriazoles.

Keywords: arylhydrazones, amidoximes, ketonitriles, isoxazoleamines, pyrazoles, 1,2,3-triazoleamines, fused 1,2,3-triazoles, fused pyridines

The synthetic potentialities of 2-arylhydrazononitriles have recently been reviewed.1 Elnagdi et al. showed that these readily obtainable²⁻⁴ versatile starting materials are excellent precursors to pyrazolamines, 1,2,4-triazolamines,⁵ 1,2,3triazolamines,⁶ isoxazol-3-amines⁷ and pyridazin-6-imines.⁸ Recently Elnagdi et al. have reported that 2-phenylhydrazonowith 3-oxobutanenitrile 1a reacted hydroxylamine hydrochloride in ethanolic sodium acetate to yield amidoxime 2a that cyclised readily into 4-acetyl-2-phenyl-1,2,3-triazol-5amine 3a upon reflux in DMF in presence of piperidine.9

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Results and discussion

As part of this work, we subjected 1b to hydroxylamine hydrochloride in ethanolic sodium acetate. Unexpectedly, a condensation product with elimination of water was obtained. We had expected that an intermediately formed amidoxime 2b would cyclise readily to 3b in a way similar to that reported recently from our laboratories. 10 However, ¹³C NMR of the reaction product indicated that this was not the case, as it indicated the absence of the carbonyl carbon in the range $\delta = 180-200$ ppm. We therefore considered the formation of isomeric isoxazoles 4b and/or 5b. While 4b can be produced via cyclisation of the initially formed 2b, formation of 5b can take place only via intermediacy of the initial product of condensation of the keto carbonyl of 1b with hydroxylamine, yielding 6b, that could then cyclise to **5b** (Scheme 1).

The subsequent chemical behaviour of the reaction product led us to conclude that 4b is the correct structure. Thus, when it was heated in DMF in the presence of piperidine it rearranged readily into 3b, which reacted readily with malononitrile to yield the triazolo[4,5-*b*]pyridine 7b (Scheme 1).

Scheme 1 Reagents: i, NH2OH.HCI, EtOH, NaOAc; ii, DMF/piperidine; iii, CH2(CN)2, EtOH

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Scheme 2 Reagents: i, Ac₂O; ii, reflux, DMF; iii, AcOH/Ac₂O.

Scheme 3 Reagents: i, Ac₂O/AcOH; ii, DMFDMA, xylene; iii, N₂H₄.H₂O/EtOH; iv, DMF, piperidine.

To our knowledge, this is the first reported rearrangement of arylazoisoxazol-3-amines into 1,2,3-triazole derivatives. In contrast to 1b, compound 1c reacted with hydroxylamine hydrochloride in ethanolic sodium acetate yielding 2c that cyclised into 3c on refluxing in DMF in the presence of piperidine. The latter also condensed with malononitrile, yielding 7c (Scheme 1).

In an attempt to cyclise 2c to the 1,2,3-triazole derivative 3c via refluxing in acetic anhydride, an acylated product was formed. Initially we thought that this product is acetyl derivatives of 3c (compounds 8c). However, 8c proved to be different from the product of reaction 2c with acetic anhydride. The latter was thus assigned structure 9c. Heating 9c in DMF gives 3c. We believe that initially a Katritzky-Boulton¹¹ rearrangement of 9c into 8c occurred and that 8c was then deacetylated under the reaction conditions.

We have investigated the potential utility of 8b,c for synthesis of azolyl-1,2,3-triazoles as well as 1,2,3triazoloazines. Refluxing 8b,c with dimethylformamide dimethylacetal (DMFDMA) afforded the trans-enaminones 10b,c based on their ¹H NMR. Thus, the ¹H NMR of 10b revealed two olefinic protons at $\delta = 5.85$ and 7.13 ppm with the coupling constant J = 12 Hz. A smaller J would be expected for cis olefinic protons.

Compounds 10b,c were refluxed with hydrazine hydrate in ethanol to yield the pyrazolyl-1,2,3-triazole 11b,c. Refluxing 10b in DMF in the presence of piperidine gave 1,2,3-triazolo-[4,5-*b*]pyridine **12b** (Scheme 3).

Coupling of 10b,c with benzenediazonium chloride gave 15b,c via intermediacy of the nonisolable 13b,c, or their hydrolysis products 14b,c. Enaminones are known to react readily with aromatic diazonium salts to yield arylazo (or arylhydrazono) derivatives. 12 This is thus a further application of this reaction.

Although 15b,c can also, at least in theory, exist as 16 or 17 or other tautomers, ¹³C NMR as well as NOE difference experiment supported the proposed hydrazone structure 15b,c (Scheme 4).

Scheme 4 Reagents: i, PhN₂+Cl⁻; ii, (deacetylation).

Experimental

IR spectra were recorded in KBr and were determined on a Perkin-Elmer 2000 FT-IR system. NMR spectra were determined on a Bruker DPX spectrometer, at 400 MHz for ¹H NMR and 100 MHz for ¹³C NMR, in CDCl₃ or DMSO-d₆ as solvent and TMS as internal standard; chemical shifts are reported in δ (ppm). Mass spectra were measured on VG Autospec QMS 30 and MS 9 (AEI) spectrometers, with EI 70 eV. Elemental analyses were measured using a LEO CHNS-932 Elemental Analyser in the Analytical Laboratory of Kuwait University.

Synthesis of hydrazono-nitriles 1b,c: general procedure

A cold solution of aryldiazonium salt (0.01 mol) was prepared by adding sodium nitrite (0.7 g) in H₂O (5 ml) to a cold solution of an arylamine hydrochloride (0.01 mol of arylamine in 5 ml concentrated HC1) with stirring. The resulting aryldiazonium salt solution was then dropped into a cold solution of 3-aminocrotononitrile (0.82 g, 0.01 mol) in ethanol (10 ml) and subsequently sodium acetate (2 g) was added. The reaction mixture was then kept at room temperature for 1 h. The resulting solid product which separated was filtered off and dissolved in acetic acid (20 ml). Concentrated hydrochloric acid (12 ml) was added, and the mixture was refluxed for 15 min, allowed to cool, and then poured into ice-water. A yellow solid was obtained, which was isolated and recrystallised from ethanol.

2-[2-(4-Methoxyphenyl)hydrazono]-3-oxobutanenitrile (1b): Yellow plates (90%), m.p. 130–132°C. IR: v_{max} 3198 (NH), 2213 (CN), 1667 cm⁻¹ (CO). NMR: $\delta_{\rm H}$ (CDCl₃) 2.5 (s, 3H, CH₃), 3.8 (s, 3H, OCH₃), 6.9 (d, 2H, J=8 Hz, Ar–H), 7.3 (d, 2H, J=8 Hz, Ar– H), 9.6 (s, 1H, NH, D_2O exchangeable); δ_C (DMSO- d_6) 193.5 (CO), 137, 7.5 (3, 111, 141, D_2) exchangeaute), 0c (DMSO- Q_6) 193.5 (CO), 157.9, 136.4, 130.3, 119.1 (2CH), 115.6 (2CH), 112.2, 56.2, 25.5 (CH₃). MS: m/z (%) 217 (M⁺, 98), 122 (65), 107 (80), 77 (50). Anal. Calcd for $C_{11}H_{11}N_3O_2$ (217.2): C, 60.8; H, 5.1; N, 19.3. Found: C, (2.6), 14.4 (2.1), 10.222 60.6; H, 4.9; N, 19.3%

2-[2-(4-Chlorophenyl)hydrazono]-3-oxobutanenitrile (1c): Yellow J = 8 Hz, Ar–H), 7.5 (d, 2H, J = 8 Hz, Ar–H), 12.3 (s, 1H, NH, D₂Oexchangeable); δ_C 193.5 (CO), 141.9, 130.3 (2CH), 129.9, 118.8 (2CH), 115.2, 111.8 56.2, 25.5 (CH₃). MS: m/z (%) 221 (M⁺, 100), 139 (40), 111 (85), 75 (15). Anal. Calcd for C₁₀H₈ClN₃O (221.6): C, 54.1; H, 3.6; N, 18.9. Found C, 54.0; H, 3.6; N, 19.0%.

2-[2-(4-Chlorophenyl)hydrazono]-N-hydroxy-3-oxobutanamidine (2c) The nitrile 1c (2.21 g, 0.01 mol), hydroxylamine hydrochloride (0.69 g; 0.01 mol), and sodium acetate (0.82 g, 0.01 mol) in ethanol (20 ml) were refluxed for 4 h. After cooling, the mixture was poured onto ice-water. The solid which separated was collected by filtration and crystallised from ethanol to give yellow needles (yield 75%), m.p. 207–208°C. IR: v_{max} 3478 (OH), 3340, 3285 (NH₂), 3155 (NH), 1697 cm⁻¹ (CO). NMR: $\delta_{\rm H}$ (CDCl₃) 2.5 (s, 3H, CH₃), 6.1 (br, 1H, OH, D₂O-exchanged), 6.5 (br, 2H, NH₂, D₂O-exchanged), 7.2 (d, 2H, *J* = 8 Hz, Ar–H) 7.3 (d, 2H, *J* = 8 Hz, Ar–H), 13.4 (s, 1H, NH, D₂O-exchanged). exchanged); $\delta_{\rm C}$ (DMSO- d_6) 199.4 (CO), 152.1, 142.4, 130.5 (2CH), 128.4, 127.8, 117.1 (2CH), 27.4 (CH₃). MS: *m/z* (%) 254 (M⁺, 95), 237 (80), 221 (40), 195 (10), 141 (20), 127 (60), 111 (40), 90 (15), 69 (45). Anal. Calcd for C₁₀H₁₁ClN₄O₂ (254.6): C, 47.1; H, 4.3; N, 22.0. Found C, 47.5; H, 4.2; N, 22.0%

1-(5-Amino-2-(4-chlorophenyl)-2H-1,2,3-triazol-4-yl)ethanone (3c) The amidoxime 2c (2.54 g, 0.01 mol) was refluxed for 6-8 h in N, N-dimethylformamide (10 ml) containing a few drops of piperidine. The solution was cooled and then poured onto ice-water. The solid which separated was collected by filtration and crystallised from petroleum ether (b.p. 60-80°C) to give yellow granules (66%), m.p. performed enter (c.p. 60–80) to give yellow granters (CO), in.p. $117-118^{\circ}\text{C}$. IR: v_{max} 3479, 3384 (NH₂), 1676 cm⁻¹ (CO). NMR (CDCl₃): δ_{H} 2.6 (s, 3H, CH₃), 5.1 (s, 2H, NH₂, D₂O-exchanged), 7.4 (d, 2H, J=8 Hz, Ar–H) 7.9 (d, 2H, J=8 Hz, Ar–H); δ_{C} 194.6 (CO), 154.6, 138.3, 133.9, 130.7 (2CH), 120.7 (2CH), 117.5, 27.2 (CH₃). MS: m/z (%) 236 (M⁺, 100), 194 (15), 125 (10), 111 (65), 90 (20), 75 (10). Anal. Calcd for $C_{10}H_9CIN_4O$ (236.6): C, 50.8; H, 3.8; N, 23.6. Found C, 51.2; H, 3.9; N, 23.3%.

4-(4-Methoxyphenyl)-5-methylisoxazol-3-amine (4b)

The nitrile **1b** (2.17 g, 0.01 mol), hydroxylamine hydrochloride (0.69 g; 0.01 mol), and sodium acetate (0.82 g, 0.01 mol) were heated to reflux for 2 h in ethanol (20 ml). After cooling, the mixture was poured into ice-water. The solid which separated was collected by filtration and crystallised from petroleum ether (60-80°C) to give yellow needles (83%), m.p. $165-166^{\circ}$ C. IR: v_{max} 3430, 3274 cm⁻¹ (NH₂). NMR (CDCl₃): δ_{H} 2.7 (s, 3H, CH₃), 3.9 (s, 3H, OCH₃), 5.3 (s, 2H, NH₂, D₂O-exchanged), 6.9 (d, 2H, J = 8 Hz, Ar–H), 7.7 (d, 2H, J = 8 Hz, Ar–H); $\delta_{\rm C}$ 171.6, 162.2, 156.5, 147.2, 125.4 (2CH), 118.2, 115.65 (2CH), 56.22 (CH₃), 11.9 (CH₃). MS: m/z (%) 232 (M⁺, 95), 221 (30), 189 (20), 147 (30), 135 (25), 121 (80), 107 (40), 73 (35), 55 (35). Anal. Calcd for C₁₁H₁₂N₄O₂ (232.2): C, 56.9; H, 5.2; N, 24.1. Found C, 56.8; H, 5.1; N, 23.8%

1-[5-Amino-2-(4-methoxyphenyl)-2H-1,2,3-triazol-4-yl]ethanone (3b) Compound 4b (2.32 g, 0.01 mol) in N,N-dimethylformamide (10 ml) containing a few drops of piperidine was heated under reflux for 4 h, cooled, and then poured into ice-water. The solid which separated was collected by filtration and crystallised from ethanol to give yellow plates(yield 75%), m.p. 143–145°C. IR: ν_{max} 3452, 3355 (NH₂), 1660 cm⁻¹ (CO). NMR: δ_{H} (CDCl₃) 2.6 (s, 3H, CH₃), 3.8 (s, 3H, CH_3), 6.3 (s, 2H, NH_2 , D_2O -exchanged), 7.1 (d, 2H, J = 8 Hz, Ar–H), 8.8 (d, 2H, J = 8 Hz, Ar–H); $\delta_{\rm C}$ (DMSO- $d_{\rm 6}$) 193.7 (CO), 159.7, 155.3, 133.4, 132.3, 120.7 (2CH), 115.7 (2CH), 80.1, 56.5 (CH₃), 27.6 (CH₃). MS: *m/z* (%) 232 (M⁺, 100), 189 (15), 148 (15), 121 (85). Anal. Calcd for C₁₁H₁₂N₄O₂ (232.2): C, 56.9; H, 5.2; N, 24.1. Found C, 57.1; H, 5.2; N, 24.1%.

Synthesis of triazolopyridines 7b,c, general procedure

Equimolecular amounts of compound **3b,c** (0.01 mol) and malononitrile (0.66 g, 0.01 mol) in N,N-dimethylformamide (10 ml) were treated with a few drops of piperidine. The reaction mixture was heated to reflux for 3 h. After cooling, the mixture was poured onto ice-water. The solid, so formed, was collected by filtration and crystallised from ethanol.

5-Amino-2-(4-methoxyphenyl)-7-methyl-2H-[1,2,3]triazolo[4,5b]pyridine-6-carbonitrile (7b): Yellow fibrous crystals (yield 62%), m.p. 277–278°C. IR: ν_{max} 3469, 3366 (NH₂), 2218 cm⁻¹ (CN). NMR: δ_{H} (CDCl₃) 2.9 (s, 3H, CH₃), 3.9 (s, 3H, CH₃), 5.4 (s, 2H, NH₂, D₂Oexchanged), 7.1 (d, 2H, J = 8 Hz, Ar–H), 8.2 (d, 2H, J = 8 Hz, Ar–H); $\delta_{\rm C}$ (DMSO- $d_{\rm 6}$) 160.9, 159.3, 157.0, 149.2, 133.7, 133.4, 122.2 (2CH), 116.3, 115.9 (2CH), 97.5, 56.67 (OCH₃), 17.17 (CH₃). MS: m/z (%) 280 (M $^+$, 80), 265 (20), 232 (40), 221 (50), 207 (20), 147 (20), 121 (30), 97 (40), 81 (50), 69 (95), 57 (80). Anal. Calcd for $C_{14}H_{12}N_6O$ (280.3): C, 59.9; H, 4.3; N, 29.9. Found C, 59.9; H, 4.3; N, 29.5%.

5-Amino-2-(4-chlorophenyl)-7-methyl-2H-[1,2,3]triazolo[4,5-b]pyridine-6-carbonitrile (7c): Yellow crystals (yield 58%), m.p. 289–290°C. IR: v_{max} 3390, 3386 (NH₂), 2230 cm⁻¹ (CN). NMR: δ_{H} (CDCl₃) 2.9 (s, 3H, CH₃), 5.5 (s, 2H, NH₂, D₂O-exchanged), 7.5 (d, 2H, J=8 Hz, Ar–H), 8.2 (d, 2H, J=8 Hz, Ar–H). δ_{C} (DMSO- d_{6}) 159.4, 157.1, 149.5, 138.9, 134.6, 134.1, 130.8 (2CH), 122.2 (2CH), 116.1, 98.6, 17.2 (CH₃). MS: m/z (%) 284 (M⁺, 65), 235 (95), 218 (50), 193 (15), 138 (30), 124 (75), 110 (80), 90 (30), 69 (35). Anal. Calcd for C₁₃H₉ClN₆ (284.7): C, 54.8; H, 3.1; N, 29.5. Found C, 54.6; H, 3.2; N, 29.2%.

Synthesis of triazoles 8b,c: general procedure

Equimolar amounts of amine 3b,c (0.01 mol) and acetic anhydride (1.02 g, 0.01 mol) in glacial acetic acid (20 ml) were heated to reflux for 2 h. After cooling, the reaction mixture was poured into ice-water. The solid so formed was collected by filtration and crystallised from petroleum ether (b.p. 60-80°C) to give yellow crystals.

N-[5-Acetyl-2-(4-methoxyphenyl)-2H-1,2,3-triazol-4-yl]acetamide (8b): Yellow granules (70%), m.p. 126–127°C. IR: v_{max} 3321 (NH), 1716 (CO), 1670 cm⁻¹ (CO). NMR (DMSO-d₆): δ_H 2.1 (s, 3H, CH₃), 2.5 (s, 3H, CH₃), 3.38 (s, 3H, OCH₃), 7.1 (d, 2H, J = 8 Hz, Ar–H), 7.9 (d, 2H, J = 8 Hz, Ar–H), 10.1 (br, 1H, NH, D₂O-exchanged); $\delta_{\rm C}$ 192 (CO), 169.4 (CO), 145.4, 138.9, 133.8, 130.9 (2CH), 130.7, 121.5 (2CH), 56.8 (OCH₃), 28.8 (CH₃), 24.2 (CH₃). MS: m/z (%) 274 (M⁺, 60), 232 (100), 217 (10), 189 (25), 135 (15), 121 (40), 107 (25), 92 (15), 77 (15), 64 (10). Anal. Calcd for C₁₃H₁₄N₄O₃ (274.3): C, 56.9; H, 5.1; N, 20.4. Found C, 56.8; H, 5.2; N, 20.3%. N-[5-Acetyl-2-(4-chlorophenyl)-2H-1,2,3-triazol-4-yl]acetamide

(8c): Yellow solid (63%), m.p. 160–162°C. IR: v_{max} 3287 (NH), 1693 (CO), 1674 cm⁻¹ (CO). NMR (DMSO- d_6): $\delta_{\rm H}$ 2.1 (s, 3H, CH₃), 2.6 (s, 3H, CH₃), 7.6 (d, 2H, J = 8 Hz, Ar–H), 8.1 (d, 2H, J = 8 Hz, Ar– H), 10.2 (br, 1H, NH, D_2O -exchanged); δ_C 192 (CO), 169.4 (CO), 145.4, 138.2, 133.9, 130.8 (2CH), 125.3, 121.1 (2CH), 28.7 (CH₃), 19.5 (CH₃). MS: m/z (%) 278 (M⁺, 30), 236 (95), 194 (25). Anal. Calcd for $C_{12}H_{11}CIN_4O_2$ (278.7): C, 51.7; H, 3.9; N, 20.1. Found C, 51.6; H, 4.0; N, 19.0%

1-[2-(4-Chlorophenyl)hydrazono]-1-(5-methyl-1,2,4-oxadiazol-3yl)propan-2-one (9c)

Equimolecular amounts of the amidoxime 2c (2.54 g, 0.01 mol) and acetic anhydride (1.02 g, 0.01 mol) in glacial acetic acid (20 ml) was Synthesis of the enaminoketones 10b,c, general procedure

The acetmidotriazole **8b,c** (0.01 mol) and N,N-dimethylformamide dimethylacetal (1.19 g, 0.01 mol) were heated to reflux for 6 h in xylene (20 ml). The reaction mixture was evaporated under reduced pressure to yield a crude product which from toluene formed dark yellow crystals.

N-[5-[3-(Dimethylamino)acryloyl]-2-(4-methoxyphenyl)]-2H-1,2,3-triazol-4-yl]acetamide (10b): Yellow granules (70%), m.p. 173–175°C. IR ν_{max} 3254 (NH), 1681 (CO), 1639 cm⁻¹ (CO). NMR: δ_H (DMSO-d₆) 2.1 (s, 3H, CH₃), 2.9 (s, 3H, NCH₃), 3.2 (s, 3H, NCH₃), 3.8 (s, 3H, OCH₃), 5.8 (d, 1H, J = 12 Hz, CH), 7.1 (d, J = 8 Hz, 2H, Ar-H), 7.8 (d, 1H, J = 12, CH), 7.9 (d, J = 8, 2H, Ar-H), 1.1 (x, 1H, NH, D, O avelaged): δ_A (CDC), 1864 (CO) Ar-H), 10.1 (br, 1H, NH, D₂O-exchanged); δ_C (CDCl₃) 186.4 (CO), 167.3 (CO), 155.5, 145.5, 142.0, 134.9, 130.9 (2CH), 128.3, 121.5 (2CH), 92.4, 56.8 (OCH₃), 46.0 (NCH₃), 38.1 (NCH₃), 25.1 (CH₃). MS: m/z (%) 329 (M⁺, 95), 312 (25), 285 (30), 243 (80), 121 (10), 98 (40), 70 (15). Anal. Calcd for $C_{16}H_{19}N_5O_3$ (329.4): C, 58.3; H, 5.8; N, 21.2. Found C, 58.1; H, 5.8; N, 21.3%.

N-[2-(4-Chlorophenyl)-5-(3-(dimethylamino)acryloyl)-2H-1,2,3triazol-4-yl]acetamide (10c): Yellow crystals (65%), m.p. 186triazoi-4-yiJacetamide (10c): Yellow crystals (65%), m.p. 186–188°C. IR: v_{max} 3265 (NH), 1686 (CO), 1637 cm⁻¹ (CO). NMR: δ_{H} (DMSO- d_{6}) 2.1 (s, 3H, CH₃), 2.9 (s, 3H, NCH₃), 3.3 (s, 3H, NCH₃), 5.8 (d, 1H, J = 12 Hz, CH), 7.6 (d, J = 8 Hz, 2H, Ar–H), 7.8 (d, 1H, J = 12 Hz, CH), 8.1 (d, J = 8 Hz, 2H, Ar–H), 10.2 (br, 1H, NH, D₂O-exchanged); δ_{C} (CDCl₃): 180.1 (CO), 167.3 (CO), 155.5, 146.3, 142.0, 138.4, 132.9, 130.5 (2CH), 120.9 (2CH), 91.9, 45.6 (CH₂), 38.1 (CH₂), 24.5 (CH₂), MS: $\frac{1}{12}$ (CP₂), 212.2 (M+ 90.216) 45.6 (CH₃), 38.1 (CH₃), 24,5 (CH₃). MS: m/z (%) 333 (M⁺, 80), 316 (40), 292 (90), 263 (40), 247 (80), 221 (40), 111 (20), 98 (95), 70 (20). Anal. Calcd for $C_{15}H_{16}ClN_5O_2$ (333.7): C, 53.9; H, 4.8; N, 20.9. Found C, 53.9; H, 4.8; N, 20.7%.

Synthesis of pyrazolyltriazoles 11b,c, general procedure

Equimolecular amounts of 10b,c (0.01 mol) and hydrazine hydrate (0.5 g, 0.01 mol) in N,N-dimethylformamide (20 ml) were refluxed for 5 h. After cooling, the reaction mixture was poured into ice-water. The solid so formed was collected by filtration and crystallised from

2-(4-Methoxyphenyl)-5-(1H-pyrazol-3-yl)-2H-1,2,3-triazol-4amine (11b): Yellow granules (70%), m.p. 149–150°C. IR: v_{max} 3433, 3387 (NH₂), 3313 cm⁻¹ (NH). NMR: $\delta_{\rm H}$ (CDCl₃) 3.8 (s, 3H, OCH₃), 4.8 (br, 2H, NH₂, D₂O-exchanged), 6.8 (d, 1H, J = 8 Hz, CH), 6.9 (d, 2H, J = 8 Hz, Ar–H), 7.6 (d, 1H, J = 8 Hz, CH), 7.9 (d, 2H, J = 8 Hz, Ar–H), 10.3 (br, 1H, NH, D₂O-exchanged); $\delta_{\rm C}$ (DMSO- $d_{\rm 6}$) 158.2, 151.7, 134.2, 130.7, 122.2, 119.2 (2CH), 115.7, 115.6 (2CH), 102.8, 56.4 (CH₃). MS: m/z (%) 256 (M⁺, 100), 241 (30), 121 (20), 107 (20), 92 (20), 77 (20). Anal. Calcd for $C_{12}H_{12}N_6O$ (256.3): C, 56.2; H, 4.7; N, 32.7. Found C, 55.9; H, 4.9; N, 32.5%.

N-[2-(4-Chlorophenyl)-5-(1H-pyrazol-3-yl)-2H-1,2,3-triazol-4yl)acetamide (11c): Yellow crystals (68%), m.p. 251–252°C. IR v_{max} 3258 (NH), 3200 (NH), 1668 cm⁻¹ (CO). NMR: $\delta_{\rm H}$ (CDCl₃) 2.1 (s, 3H, CH₃), 6.6 (d, 1H, J = 8 Hz, CH), 7.6 (d, 2H, J = 8 Hz, Ar–H), 7.8 (d, 1H, *J* = 8 Hz, CH), 8.0 (d, 2H, *J* = 8 Hz, Ar–H), 10.0 (br, 1H, NH, D_2O -exchanged), 13.2 (br, 1H, NH, D_2O -exchanged); δ_C (DMSO- d_6) 170.0 (CO), 142.8, 142.1, 140.8, 138.7, 132.6, 130.7 (2CH), 120.5

(2CH), 105.9, 104.5, 19.5 (CH₃). MS: m/z (%) 302 (M⁺, 70), 260 (95), 125 (40), 111 (25), 94 (30), 75 (15). Anal. Calcd for C₁₃H₁₁ClN₆O (302.7): C, 51.5; H, 3.6; N, 27.7. found C, 51.6; H, 3.6; N, 27.7%.

Synthesis of the triazolopyridines 15b,c, general procedure

Benzenediazonium chloride solution was prepared by the standard literature procedure from aniline (0.93 g, 0.01 mol), concentrated hydrochloric acid, and sodium nitrite. Compound 10b,c (0.01 mol) in ethanol (50 ml) was stirred with sodium acetate (5 g), and the diazonium salt solution was added gradually to the mixture. After the addition was complete, the reaction mixture was kept at room temperature for 1 h. The solid which separated was collected by filtration and crystallised from ethanol to give a dark yellow powder.

2-(4-Methoxyphenyl)-6-(2-phenylhydrazono)-2H-[1,2,3] triazolo[4,5-b]pyridin-7(6H)-one (15b): Yellow solid (79%), m.p. 253–255°C. IR: v_{max} 3435 (NH), 1626 cm⁻¹ (CO). NMR: δ_{H} (DMSO d_6) 3.8 (s, 3H, OCH₃), 7.1 (d, 2H, J = 8 Hz, Ar–H), 7.4-7.8 (m, 5H, Ar-H), 8.1 (d, 2H, J = 8 Hz, Ar-H), 8.4 (s, 1H, CH), 15.8 (br, 1H, NH, D_2O -exchanged); δ_C (CDCl₃) 169.3 (CO), 161.0, 159.4, 141.4, 140.2, 135.8, 132.9, 130.4 (2CH), 130.0 (2CH), 130, 129.7, 119.3 (2CH), 115.0 (2CH), 56.6 (CH₃). MS: m/z (%) 346 (M⁺, 100), 269 (15), 241 (75), 158 (25), 107 (25), 72 (25), 77 (70). Anal. Calcd for C₁₈H₁₄N₆O₂ (346.35): C, 62.4 H, 4.0; N, 24.2. Found C, 62.2; H, 3.9; N, 24.4%.

((Z)-2-(4-Chlorophenyl)-6-(2-phenylhydrazono)-2H-[1,2,3]triazolo[4,5-b]pyridin-7(6H)-one (15c): yellow solid m.p. 192–94°C. IR: v_{max} 3355 (NH), 1629 cm⁻¹ (CO). NMR: δ_{H} (DMSO- d_{6}) 7.3–7.8 (m, 7H, Ar–H), 8.2 (d, 2H, J = 8 Hz, Ar–H), 8.5 (s, 1H, CH), 16.0 (br, 1H, NH, D₂O-exchanged); δ_C (CDCl₃) 169.4 (CO), 161.0, 159.4, 141.4, 140.2, 135.8, 132.9, 130.4 (2CH), 130.2 (2CH), 130, 129.7, 122.3 (2CH), 120.0 (2CH). MS: *m/z* (%) 350 (M⁺, 100), 273 (30), 245 (45), 210 (15), 182 (25), 162 (30), 125 (20), 111 (70), 93 (20), 77 (98). Anal. Calcd for C₁₇H₁₁ClN₆O (350.7): C, 58.2; H, 3.1; N, 23.9. Found C, 58.2; H, 3.0; N, 23.7%.

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References

- S. M. Riyadh, I. A. Abdelhamid, H. M. Ibrahim, H. M. Al-Matar and M. H. Elnagdi, Heterocycles, 2007, in press
- F. Al-Omran, M. M. Abdel Khalik, A. A. ElKhair and M. H. Elnagdi,
- N. M. Toder Khain, A. A. Erkhain and M. H. Ellingdi, Synthesis, 1997, 91-94.
 V. Kryštof, P. Cankar, I. Fryšová, J. Slouka, G. Kontopidis, P. Dzubák, M. Hajdúch, J. Srovnal, W.F. de Azevedo, M. Orság, M. Paprskárová, J. Rolcík, A. Látr, P.M. Fischer, and M. Strnad, *J. Med. Chem.*, 2006, 49, 9500.
- 4 S. A. Abdallah, N. H. Metwally, H. F. Anwar, and M. H. Elnagdi, J. Heterocycl. Chem., 2005, 42, 781
- 5 H.M. Al-Matar, S.M. Riyadh and M.H. Elnagdi, Arkivoc, 2007, xii, 53.
- S.I. Aziz, H.F. Anwar, D.H. Fleita and M.H. Elnagdi, J. Heterocycl. Chem., 2007, 44, 725
- M. H. Elnagdi, M. R. H. Elmoghayer, E. A. Hafez and H. H. Almina, J. Org. Chem., 1975, 40, 2604.
 H. Al-Awadhi, F. Al-Omran, M. H. Elnagdi, L. Infantes, C. Foces-Foces, N. Jagerovic and J. Elguero, Tetrahedron, 1995, 51, 19.
- S.A.S. Ghozlan, I.A. Abdelhamid, H.M. Ibrahim and M.H. Elnagdi, Arkivoc, 2006, xv, 53.
- 10 S. M. Al-Mousawi and M. S. Moustafa, Beilstein J. Org. Chem., 2007, 3 12
- 11 A.J. Boulton and A.R. Katritzky, Rev. Chim. (Acad. R.P. Roumaine), 1962, 7, 691.
- 12 F. Al-Omran, M.M. Abdel Khalik, A.A. ElKhair and M.H. Elnagdi, Synthesis, 1997, 91.